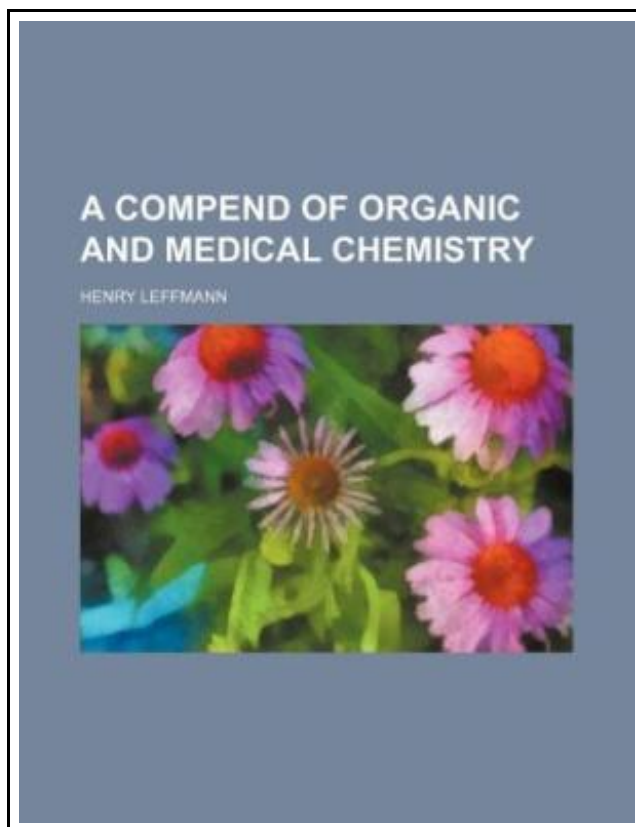


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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1884 Excerpt: . been obtained from these bodies, but only a few are important. Potassium sulphocyanate, KCNS, is a white, crystalline body, very soluble in water, producing, with ferric salts, a deep blood-red color, which is utilized as a delicate test. Ammonium sulphocyanate, NH_4CNS , is formed when hydrogen cyanide acts on ammonium sulphide, which contains some free sulphur. The reaction is $(\text{NH}_4)_2\text{S} + \text{S}_2 + 2\text{HCN} = 2\text{NH}_4\text{CNS} + \text{H}_2\text{S}$. This reaction is the basis of one of the tests for cyanogen. Ammonium cyanate, NH_4CNO . This is one of the most interesting of the derivatives of the compounds of this class, because it is identical in composition, and may be easily converted into urea, the most abundant solid constituent of the urine. It may be formed by the action of ammonium sulphate on potassium cyanate, and when heated, either in the solid or dissolved, it soon becomes converted, without change of composition, into a body from which neither cyanic acid nor ammonia can be obtained by the usual tests. This fact shows that the atoms have been rearranged. The body, in fact, has been converted into a substitution ammonium compound, and is identical, in every respect, with urea as obtained from urine. SUBSTITUTION AMMONIUMS. Ammonia, NH_3 , is always found, either free or combined, among the products of decomposition of nitrogenous matter. As a distinct substance, one of its most striking properties is its power to neutralize acids. It was natural, therefore, that, when organic bodies containing nitrogen were found to have a...



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